

School Level Examination SLE 2022



Subject Code: 3 0

Total Questions: 50

Time: 1 hour

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DO NOT OPEN THIS BOOKLET UNTIL INSTRUCTED TO DO SO

- > All questions are compulsory.
- Read the instructions on the ANSWER SHEET and fill in your NAME, CLASS and OTHER INFORMATION.
- To mark your choice of answer by darkening the circles in the ANSWER SHEET, use a BLUE/BLACK BALL PEN only.
- > You **MUST** record your answers on the **ANSWER SHEET** only.
- There are 50 MULTIPLE CHOICE QUESTIONS.
 Use the information provided to choose the BEST answer among the four possible options. On your ANSWER SHEET fill in the circle that matches your answer.
- > Marks are **NOT** deducted for incorrect answers.
- > Return the **ANSWER SHEET** to the invigilator at the end of the examination.
- You are **NOT** allowed to use a calculator.
 You may use a ruler and spare paper for rough work.



6. How many squares do you see in the given image?

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(A) 9(B) 10(C) 11(D) 12

- 7. Last weekend, I went to play in the nearby park. It was real fun! I rode my new bicycle that Mom had gifted me on my birthday.

On reaching the park, I saw that there were total 12 bicycles and tricycles. If the total number of wheels was 29, how many tricycles were there?

(A)	7	(B)	6
(C)	5	(D)	8

8. How many triangles can you count in this cat?

(A)	24	(B)	27
(C)	28	(D)	21

Here are some words translated from an artificial language.

- 'Titi poi' means deep sleep. •
- 'Kuha rosa' means start jumping. •
- 'Poi koi' means deep sea.
- Which word could mean "sleep start"? •
- (A) Titi rosa (B) Poi kuha
- (C) Choch titi (D) Poi poi
- 10. Which number can replace the question mark?

	2	3	17
	11	13	?
	7	19	5
(A) 21			(B) 23
(C) 25			(D) 1

Section B (Subject Specific)

(This section contains 25 multiple choice questions. Each question has four choices (A), (B), (C) and (D), out of which only ONE is correct.)

11. The following reaction is an example of _____

 $4 \text{ NH}_3(g) + 5O_2(g) \rightarrow 4\text{NO}(g) + 6\text{H}_2\text{O}(g)$

- displacement reaction Ι.
- redox reaction III.
- (A) Land II
- (C) I and III (D) III and IV

12. What is the ionic equation for the neutralization reaction of aqueous sodium hydroxide and dilute nitric acid?

(A) $HNO_3 + NaOH \longrightarrow NaNO_3 + H_2O$

(C) NaOH + $H^+ \longrightarrow Na^+ + H_2O$

- (B) $HNO_3 + OH^- \longrightarrow H_2O + NO^{3-}$ (D) $Na^+ + HNO_3 \longrightarrow NaNO_3 + H^+$

exothermic reaction

IV. neutralization reaction

13. Which of the following is correct general formula of saturated carboxylic acids?

(A) $C_n H_{2n-1} COOH$

- (B) C_nH_{2n}COOH
- (C) $C_n H_{2n+1} COOH$
- (D) C_nH2_{n-2}COOH

11.

(B) II and III

- 14. Which of the following furnaces are used for producing very high temperatures?
 - (A) Blast furnace
 - (C) Regeneration furnace

(B) Reverberatory furnace

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(D) Electric furnace

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15.	mber for the elements properties CANNOT be			
	(A) Size of	the ion		Lope
	(B) Size of a	atom		Atomic number
	(C) Charge	on the ion		
_	(D) Numbe	r of electron shells		
16.	The human d	ligestive system consists of		
	I. mouth,	small intestine, large intestine	١١.	gullet, stomach, anus
	III. heart, l	ung, kidney	IV.	liver
	(A) I and II		(B)	II and III
	(C) I and III		(D)	I, II and IV
17.	Hormone whimplantation	nich suspends ovulation during pregnand of the embryo is	cy and	d prepares the uterine endometrium for the
	(A) oestrog	en	(B)	progesterone
	(C) oxytoci	n	(D)	All of these
18.	Which of the	following is not a STD (Sexually Transmit	ted D	isease)?
	(A) Gonorr	hoea	(B)	Syphilis
	(C) AIDS		(D)	Hepatitis
19.	Select the sta	atements that describe the characteristic	s of ge	enes.
	I. Genes a	are specific sequence of bases in a DNA n	noleci	ule.
	II. A gene	codes for a protein.		
	III. In indiv	iduals of a given species, a specific gene	is loca	ated on a particular chromosome.
	IV. Each ch	romosome has only one gene.		
	(A) I and II		(B)	II and III
	(C) III and I	V	(D)	I, II and III
20.	The diagram given below represents a ray of light striking a plane mirror. What is the angle of incidence if the total angle between incident and reflected rays is 80°?			
	(A) 40°		(B)	50°
	(C) 80°		(D)	60°
21.	The physics t	eacher uses bifocal cylinderical lens in his	s spec	tacles. He is suffering from
	(A) myopia	and astigmatism	(B)	astigmatism and hypermetropia
	(C) myopia	and hypermetropia	(D)	myopia, hypermetropia and astigmatism
22.	A thin fuse w to	A thin fuse wire of 3A fuse is connected to a plug which is connected to a table lamp. The fuse is needed to		
	(A) increase	e resistance of the circuit	(B)	decrease resistance
	(C) protect	the wire from overheating	(D)	reduce voltage across the lamp

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- 23. Choose the correct statements.
 - I. Fleming's right hand thumb rule is used to know the direction of induced current.
 - II. The right hand thumb rule is used to find the direction of magnetic field.
 - III. DC flows in one direction, AC reverses its direction.
 - IV. In India, the AC changes its direction 50 times per second.
 - (A) I and II (B) II and III
 - (C) III and IV (D) I, II and III

24. The gas released from the supersonic jet aeroplane which depletes ozone layer is ______.

- (A) CFC (B) NO
- (C) NO_2 (D) SO_2

25. If the tertiary consumer has 10kJ of energy, what will be energy of producer?

- (A) 1000 kJ (B) 100 kJ
- (C) 10000 kJ (D) 100,000 kJ

26. Environment can be protected by using integrated pest management (without or minimum use of pesticide).

- I. Improved sanitation practice
- III. Soil conditioning
- (A) I and II
- (C) I and III

IV. Biological control(B) II and III

II. Crop rotation

(D) I, II, III and IV

(D) Annealing

27. Which is the best suitable reason for choosing below mentioned sources of water?

- (A) Ground water: clean and easily usable
- (B) Underground water: does not evaporate, recharges well and clean
- (C) Ground water: contaminated
- (D) Underground water: does contain bleaching agents such as chlorine

28. When we heat some water on a stone, its temperature increases. In this case, ______.

- (A) heat energy is converted into thermal energy
- (B) thermal energy is converted into heat energy
- (C) mechanical energy is converted into thermal energy
- (D) thermal energy is converted into mechanical energy

29. The process of depositing molten Zn layer on iron articles is called ______

- (A) Anodising (B) Electroplating
- (C) Galvanisation
- 30. Ethanol on oxidation gives _____
 - (A) Ethane (B) Formalin
 - (C) Ethanoic acid (D) Methane

31. Arrange the following elements in the order of their decreasing metallic character.

- (A) Cl > Si > Al > Mg > Na (B) Na > Mg > Al > Si > Cl
- (C) Na > Al > Mg > Cl > Si (D) Al > Na > Si > Ca > Mg



- 32. Flowers like tulip, crocus open at high temperature, but close with fall of temperature. Which of the following movements does it involve?
 - (A) Photonasty (C) Seismonasty

- (B) Nyctinasty (D) Thermonasty
- 33. Which one of the given fruits develops without fertilisation?

- (1) Apple
- (III) Banana
- (A) (I) and (III) (C) (II) and (IV)

- (II) Mango
- (IV) Strawberry
- (B) Only (I) (D) Only (III)
- 34. Some dinosaurs had feathers although they could not fly but birds have feathers that help them to fly. In the context of evolution, this means that .
 - (A) reptiles have evolved from birds
 - (B) there is no evolutionary connection between reptiles and birds
 - (C) feathers are homologous structures in both birds and organisms
 - (D) birds have evolved from reptiles
- 35. "A body having large mass only shows its particle nature". Choose the correct option regarding this statement.
 - (A) According to de Broglie relation, wavelength is inversely proportional to mass of the body.
 - (B) A wavelength is directly proportional to Planck's constant.
 - (C) Planck's constant is independent variable.
 - (D) Statement is incorrect.

Section C (Competency Based)

(This section contains 15 multiple choice questions. Each question has four choices (A), (B), (C) and (D), out of which TWO are correct.)

- 36. Formalin is an aqueous solution of .
 - (A) Formic acid
 - (C) Methanal

- (B) Formaldehyde
- (D) Acetaldehyde

- 37. Sodium carbonate is .
 - (A) acidic
 - (C) neutral

- (D) salt of strong base and weak acid
- 38. What happens when a solution of an acid is mixed with a solution of base in a test tube?
 - (A) The temperature of the solution increases. (B) The temperature of the solution decreases.
 - (D) Salt formation takes place.
 - (C) The temperature of the solution remains same.
- 39. Aqua regia can dissolve _____
 - (A) Ag (B) Pb (C) Pt (D) Au
- 40. In which of the following examples is a convex mirror used?
 - (A) Solar furnace
 - (C) Street lamps

- (B) Rear view mirror in vehicles
- (D) Headlight of a car

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- (B) basic

(A) washing soda bleaching powder (B) (C) baking soda (D) slaked lime 42. Which of the following are enzymes used in digestion? Pepsin (A) (B) Trypsin (C) Lysozyme (D) Urease 43. Which of the following are NOT true for biogas? (A) Biogas contains up to 75% methane. (B) Biogas burns with minimal smoke leaving residues like ash in wood, etc. (C) Biogas is used for lighting. (D) The slurry left behind is a waste and discarded after biogas is obtained. 44. Which of the following phenomenon occurs when a small amount of acid is added to water? (A) Ionisation (B) Neutralisation Dilution (D) Salt formation (C) (A) Red light has highest wavelength. (B) Red light moves fastest. (C) All colours of white light move with same speed. (D) Yellow light moves slowest. (A) Western Ghats (B) Eastern Himalayas (C) Deccan Plateau (D) Gangetic plain true about it? (A) It is an exothermic reaction. (B) It is an endothermic reaction. (C) The pH of resulting solution is more than 7. (D) The pH of resulting solution is less than 7. 48. Which of the following substances can be added to denature ethanol? (A) Methanol (B) Pyridine (C) Butanoic acid (D) Ethanoic acid (A) Waste disposal is easy. (B) A small amount of nulcear fuel releases large amount of energy. (C) Once the nuclear fuel is loaded, nuclear reactor can operate for 2 to 3 years. (D) It does not pollute environment and is cost effective. (A) Eye donors can belong to any age group or sex. (B) Eye must be removed within one hour after death.

- (C) The removal of eye takes one hour.
- (D) Persons died because of infections or communicable diseases such as AIDS, rabies, etc. cannot donate eyes.

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- 45. Which of the following statements are correct?

46. Which of the following regions of our country are known for their rich biodiversity?

47. Solid Calcium oxide reacts vigorously with water forming calcium hydroxide. Which of the following are

- 49. Which are advantages of nuclear reactor?
- 50. Which of the statements are true regarding eye donation?

41. Common salt besides being used in kitchen can also be used as the raw material for making

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